

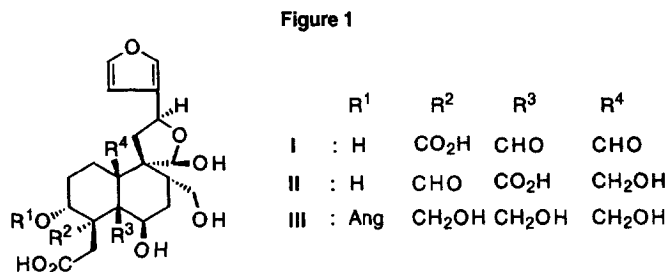
## A New Stereocontrolled Access to Angularly Disubstituted *cis*-Decalins via Tandem Radical Ring Expansion and Cyclization

Hideo Nemoto, Motohiro Shiraki, Natsuko Yamada, Naomi Raku,  
 and Keiichiro Fukumoto\*

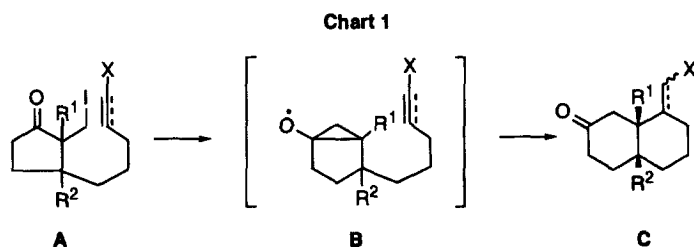
Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980-77, Japan

**Abstract:** The complete stereocontrolled access to angularly disubstituted *cis*-decalins based on the tandem radical ring expansion and cyclization process of  $\alpha$ -iodomethylcyclopentanones **1** and **6** is described. The compounds **2–5** and **7–13** thus prepared could be versatile intermediates for the synthesis of biologically important compounds. Copyright © 1996 Elsevier Science Ltd

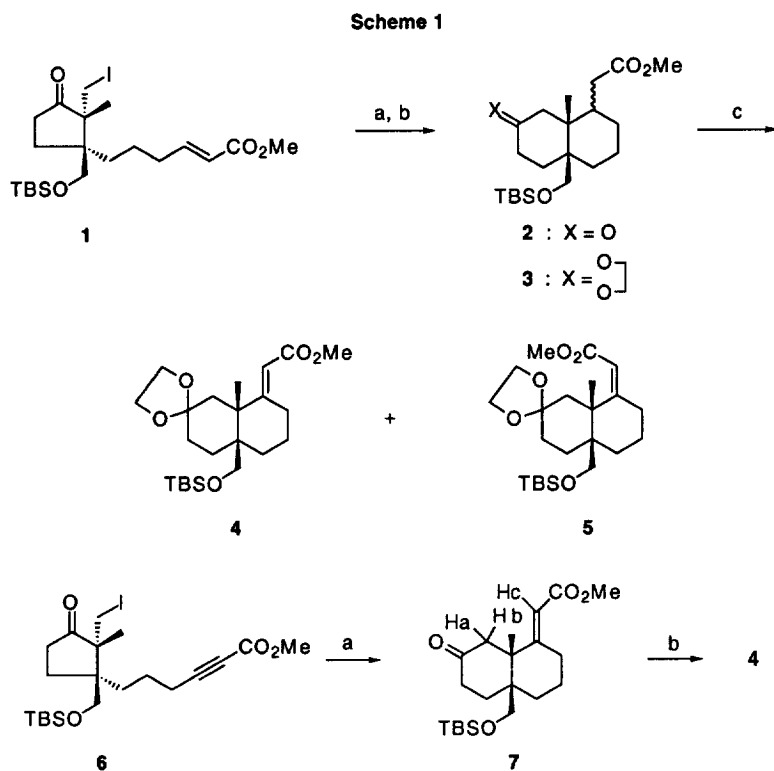
Angularly disubstituted *cis*-decalins constitute basic framework of many types of biologically important compounds<sup>1</sup> including neo-clerodane diterpenes musabalsisianes **A**, **B** and **C** (I–III)<sup>2</sup> isolated from the seeds of *Musa balbisiana* which show amoebicidal activity *in vitro* (Figure 1).



We now communicate a novel stereocontrolled synthesis of angularly disubstituted *cis*-decalins **C**. This approach is stemmed from the tandem radical ring expansion and cyclization process<sup>3</sup> of  $\alpha$ -iodomethyl substituted cyclopentanones **A** via cyclopropyl alkoxy radical **B**, by taking advantage of easy access to **A** which was recently developed by us<sup>4</sup> (Chart 1).



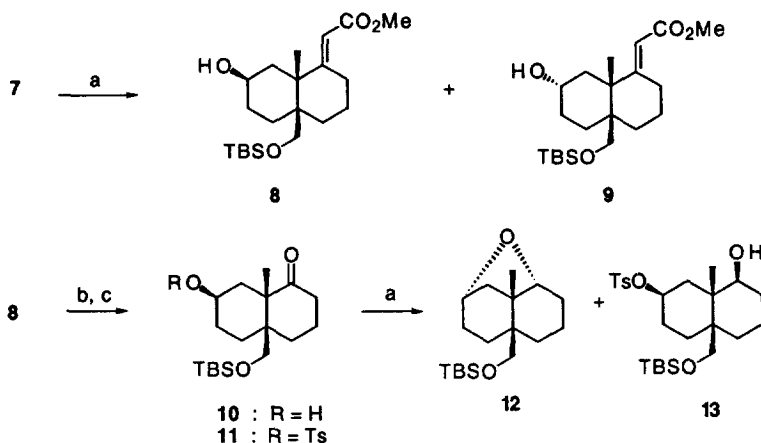
The tandem reaction of **1**<sup>4</sup> was effectively proceeded under radical conditions to give the *cis*-decalins **2** (1:1 mixture of diastereomers) in 97% yield. The ketal **3** derived (85%) from **2** was converted to the easily separable mixture (1:4) of *E* (**4**)<sup>5</sup> and *Z* (**5**)<sup>5</sup> unsaturated esters in 67% yield by phenylselenenylation-oxidative elimination sequences. The acetylenic iodide **6**<sup>4</sup> was also subjected to the same conditions for **1** to afford the *cis*-decalin **7** selectively in 85% yield. On ketalization **7** furnished the ketal **4** (66%) identical with that described above (Scheme 1).



<sup>a</sup>Steps: (a) <sup>n</sup>Bu<sub>3</sub>SnH, AIBN, benzene, reflux, 5 h; (b) HOCH<sub>2</sub>CH<sub>2</sub>OH, *p*-TsOH, benzene, reflux, 9 h; (c) (1) LDA, PhSeBr, THF, -78 °C, 4.5 h; (2) 30% H<sub>2</sub>O<sub>2</sub>, NaHCO<sub>3</sub>, THF, rt, 2 h.

Although the *E*-geometry of **7**<sup>5</sup> was confirmed by the observation of definite *n*Oe between Ha and Hb [2.21 and 2.82 ppm (each 1H, each d, *J* = 15.0 Hz)], and Hc [5.68 ppm (1H, s)] in its NMR (500 MHz) spectrum the *cis* ring juncture of **7** was determined unambiguously giving the suitably functionalized *cis*-decalins for further elaboration as follows. Reduction of the ketone **7** afforded the alcohols **8** and **9** (4:1) in 69% yield, the former of which was then subjected to ozonolysis to give the keto alcohol **10** (55%). Finally, the tosylate **11** derived (81%) from **10** was reduced to furnish the tricyclic compound **12** (59%) together with the alcohol **13** (38%) confirming the *cis* ring juncture of **7** (Scheme 2).

Scheme 2



<sup>a</sup>Steps: (a) NaBH<sub>4</sub>, MeOH, 0 °C, 15 min; (b) O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C, 3 h then Me<sub>2</sub>S;  
 (c) *p*-TsCl, pyridine, DMAP, 0 °C, 24 h.

Thus, we could provide an efficient pathway to disubstituted *cis*-decalins which could be potential intermediates for the synthesis of biologically important compounds.

## References and Notes

- For naturally occurring *cis*-decalins, see: (a) Devon, T. K.; Scott, A. I. *Handbook of Naturally Occurring Compounds*, Vol. I and II, Academic Press, New York, 1972. (b) *Terpenoids and Steroids*, The Chemical Society, London, Vols. 1-12. (c) Hanson, J. R. *Nat. Prod. Rep.* **1996**, *13*, 59-71. (d) Faulkenr, D. J. *Nat. Prod. Rep.* **1996**, *13*, 75-125.
- Ali, M. *Phytochemistry* **1992**, *31*, 2173-2175.
- For reviews in this area, see: (a) Giese, B. *Radicals in Organic Synthesis: Formation of Carbon-Carbon Bonds*, Pergamon Press, New York, 1986. (b) Fossey, J.; Lefort, D.; Sorba, J. *Free Radicals in Organic Chemistry*, John Wiley & Sons, Paris, 1995. (c) Curran, D. P. *Synthesis* **1988**, 417-439 and 489-513. (d) Jasperse, C. P. Curran, D. P.; Fevig, T. L. *Chem. Rev.* **1991**, *91*, 1237-1286. (e) Porter, N. A.; Giese, B.; Curran, D. P. *Acc. Chem. Res.* **1991**, *24*, 296-306. (f) Dowd, P.; Zhang, W. *Chem. Rev.* **1993**, *93*, 2091-2115. (g) Smadja, W. *Synlett* **1994**, 1-26. For some recent studies, see: (h) Dowd, P.; Choi, S.-C. *J. Am. Chem. Soc.* **1987**, *109*, 3493-3494 and 6548-6549. (i) Beckwith, A. L. J.; O'Shea, D. M.; Gerba, S.; Westwood, S. W. *J. Chem. Soc., Chem. Commun.* **1987**, 666-667. (j) Beckwith, A. L. J.; O'Shea, D. M.; Westwood, S. W. *J. Am. Chem. Soc.* **1988**, *110*, 2565-2575. (k) Ellwood, C. W.; Pattenden, G. *Tetrahedron Lett.* **1991**, *32*, 1591-1594. (l) Nishida, M.; Ueyama, E.; Hayashi, H.; Ohtake, Y.; Yamamura, Y.; Yanaginuma, E.; Yonemitsu, O.; Nishida, A.; Kawahara, N. *J. Am. Chem. Soc.* **1994**, *116*, 6455-6456. (m) Curran, D. P.; Qi, H.; DeMello, N. C.; Lin, C.-H. *ibid.* **1994**, *116*,

- 8430-8431. (n) Molander, G. A.; Harris, C. R. *ibid.* **1995**, *117*, 3705-3716; *ibid.* **1996**, *118*, 4059-4071.
4. (a) Nemoto, H.; Shiraki, M.; Fukumoto, K. *Tetrahedron Lett.* **1995**, *36*, 8799-8802. (b) Nemoto, H.; Shiraki, M.; Fukumoto, K. *J. Org. Chem.* **1996**, *61*, 1347-1353.
5. Selected data for **4**: IR  $\nu$  max (neat)  $\text{cm}^{-1}$  1710, 1630;  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  -0.04 (3H, s), -0.03 (3H, s), 0.83 (9H, s), 1.15 (3H, s), 1.42-1.88 (10H, m), 2.08-2.19 (1H, m), 2.26 (1H, d,  $J = 14.0$  Hz), 3.22 and 3.43 (each 1H, each d,  $J = 9.8$  Hz), 3.66 (3H, s), 3.73-3.97 (4H, m), 5.72 (1H, s);  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  -5.6, -5.5, 18.2, 20.3, 21.7, 25.1, 25.8, 29.4, 30.4, 41.9, 43.1, 44.4, 50.9, 63.6, 64.5, 66.0, 109.2, 113.9, 167.7, 167.9; MS  $m/z$  424 ( $\text{M}^+$ ); HRMS ( $\text{M}^+$ ) calcd for  $\text{C}_{23}\text{H}_{40}\text{O}_5\text{Si}$  424.2645. found 424.2652. For **5**: IR  $\nu$  max (neat)  $\text{cm}^{-1}$  1710, 1640;  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  -0.02 (3H, s), -0.01 (3H, s), 0.84 (9H, s), 1.20 (3H, s), 1.41-2.04 (9H, m), 2.04-2.12 (1H, m), 2.19 (1H, d,  $J = 14.0$  Hz), 2.38-2.49 (1H, m), 3.25 and 3.56 (each 1H, each d,  $J = 9.8$  Hz), 3.67 (3H, s), 3.79-3.98 (4H, m), 5.59 (1H, s);  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  -5.5, -5.4, 18.3, 19.6, 21.9, 25.9, 26.0, 28.8, 30.4, 35.4, 41.5, 41.9, 44.2, 51.6, 63.7, 64.5, 65.8, 109.2, 117.0, 153.2, 170.0; MS  $m/z$  424 ( $\text{M}^+$ ); HRMS  $m/z$  ( $\text{M}^+$ ) calcd for  $\text{C}_{23}\text{H}_{40}\text{O}_5\text{Si}$  424.2645. found 424.2621. For **7**: IR  $\nu$  max (neat)  $\text{cm}^{-1}$  1720, 1710, 1630;  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  0.03 (6H, s), 0.87 (9H, s), 1.11 (3H, s), 1.21-1.95 (6H, m), 2.21 and 2.82 (each 1H, each d,  $J = 15.0$  Hz), 2.39 (2H, t,  $J = 7.0$  Hz), 2.91-3.05 (2H, m), 3.55 and 3.62 (each 1H, each d,  $J = 9.9$  Hz), 3.67 (3H, s), 5.68 (1H, s);  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )  $\delta$  -5.6, -5.4, 18.2, 21.8, 25.1, 25.9, 26.7, 28.1, 30.4, 37.7, 42.3, 47.4, 50.0, 51.1, 66.5, 114.5, 165.0, 167.3, 210.9; MS  $m/z$  323 ( $\text{M}^+ -57$ ); HRMS  $m/z$  ( $\text{M}^+ -57$ ) calcd for  $\text{C}_{17}\text{H}_{27}\text{O}_4\text{Si}$  323.1679. found 323.1642.

(Received in Japan 13 June 1996; revised 8 July 1996; accepted 12 July 1996)